# Quaternary Research

Horváth Erzsébet ELTE TTK FFI Természetföldrajzi Tanszék

Lektorálta: Mari László

A jegyzet megjelenését az ELTE jegyzettámogatási pályázata támogatta 2021





# QUATERNARY RESEARCH

- Horváth Erzsébet
- erzsebet.horvath@ttk.elte.hu

# QUATERNARY RESEARCH

- The main topics of the course:
  - Boundary of the Quaternary
  - Climate of the Quaternary
  - Quaternary dating methods
    - Relative chronology
    - Absolute chronology
  - Quaternary in the Carpathian-Basin

# Boundary of the Quaternary





#### Quaternary Period with the Anthropocene Epoch



Surface temperature of the Greenland ice sheet

("Last Glacial Maximum" (LGM), Older Dryas, Bølling-Allerød and Younger Dryas) (http://www.dandebat.dk/eng-klima6.htm)

• Boundary of the Quaternary

# The actual distribution of the ice cover:

- 15.000.000 km<sup>2</sup>
  - Antartctica: 13.500.000 km<sup>2</sup> (max. thickness: 4270 m),
  - Greenland: 1.800.000 km<sup>2</sup> (max. thickness: 3240 m)

The ice of Antarctica and Greenland 97% of the Earth ice surface, and 99% of the ice cover of the Earth



During the maximal distribution of the glacial: 35.000.000 km<sup>2</sup> Extra:

- N-America: 12.000.000 km<sup>2</sup>
- Eurasia: 6.000.000 km<sup>2</sup>

### Glacials and interglacials in the world

		Alps	Middle-Europe		Eastern-Europe	North-America
glacial		würm	wisztula		valdaj	wisconsin
interglacial		riss-würm	eem		mikulino	sangamon
glacial	stadial	riss II	sa	warta	moszkvai	iowa
	interstadial		а	trene	odinzovo	
	stadial	riss I	le	drente	dnyeper	illinois
interglacial		mindel-riss	holstein		lichvin	yarmouth
glacial		mindel	elster		oka	kansas
interglacial		günz-mindel	cromer		bjelovesz	afton
glacial		günz	menap		narev	nebraska
interglacial		duna-günz	waal			
glacial		duna	eburon			
interglacial		duna-biber	tegelen			
glacial		biber				



#### Ice cover in the Alps during the Pleistocene









Temperatures of the Last 10,000 Years

### **Boundary of the Quaternary:**

Pleistocene/Holocene

#### THE MEDIEVAL WARM PERIOD AND THE LITTLE ICE AGE





https://uwcm-geog.wikispaces.com/EXTREME+ENVIRONMENTS

## Vegatation cover during the Last Glacial Maximum (LGM) 20000 years before present

• Comparison of the distribution of vegetation types and ice sheet extent between (A) modern and (B) Last Glacial Maximum (LGM) (13, 14). The LGM was marked by much wider distribution of dry vegetation biotopes, including those of deserts, grasslands and savanna, semidesert and dry steppe, polar desert, and southern steppe tundra compared with the present day. [Reproduced with permission from ref. 14 (Copyright 2003, American Geophysical Union)

 https://www.semanticscholar.org/paper/Glacialdemise-and-methane's-rise-Behl/295890b12b8c07129b7b32c0e459ab9e16bb7e1c/fig ure/0











## **Pleistocene Glaciation**



## The climate of the Quaternary



## Milankovich theory and the climate changes



# More details about Milankovitch theory:

https://climate.nasa.gov/news/2948/milankovitch-orbitalcycles-and-their-role-in-earths-climate/ The whole article: https://climate.nasa.gov/news/2948/milankovitch-orbital-cycles-and-their-role-in-earths-climate/

### Changes in Eccentricity (Orbit Shape)

100,000-year cycles



\*Changes in eccentricity exaggerated so the effect can be seen. Earth's orbit shape varies between 0.0034 (almost a perfect circle) to 0.058 (slightly elliptical).

climate.nasa.gov



### Axial Precession (Wobble)

26,000-year cycles





### Earth at Winter Solstice (~Dec. 21)



https://uwcm-geog.wikispaces.com/EXTREME+ENVIRONMENTS



In: Maslin, NV. 2016



Past and future Milankovitch cycles via VSOP model • Graphic shows variations in five orbital elements: Axial tilt or obliquity ( $\varepsilon$ ). Eccentricity (e). Longitude of perihelion ( $sin(\varpi)$ ). Precession index ( $e \sin(\varpi)$ ) Precession index and obliguity control insolation at each latitude: Daily-average insolation at top of atmosphere on summer solstice () at 65° N Ocean sediment and Antarctic ice strata record ancient sea levels and temperatures: Benthic forams (57 widespread locations) Vostok ice core (Antarctica) Vertical gray line shows present (2000 CE)

https://upload.wikimedia.org/wikipedia/commons/5/53/MilankovitchCyclesOrbitandCores.png

### **Complex explanation of climate changes**

Not only astronomical reasons!

### **Other important factors:**

- Oceanic currents
- Feedback of the ice sheets
- Volcanic activity







#### COMPARISON OF MARINE AND CONTINENTAL DIVISIONS IN THE PLEISTOCENE



MIS 1 – Holocén MIS 2-5d – Würm (Weischelian) MIS 5e – RW interglac. (Eemian) MIS 6 – Riss (Saalian) MIS 11 – MR interglac. (Holstein)





https://www.thegwpf.com/understanding-climate-cycles-or-how-to-avoid-climate-panics/

# Temperature changes in the Bernese Alps in the last 500 years





Changes of the ice mass of the Aletsch glacier (Western Alps, Switzerland) during the last 2000 years







1856

1987





### Aletsch glacier (CH)



- Lenght: 23 km
- Surface: 86 km<sup>2</sup>
- Waterflow in summer: 80 m<sup>3</sup>/sec
- Average retreat: 30m/year (sometimes: 90m/year)

### Aletsch glacier from the Eggishorn



### Retreat of the Aletsch glacier







# Aletschwald (330ha) (Protected since 1933, Pro Natura site)









### Retreat of the Pasterze glacier (Eastern-Alps, Austria)



2000


## **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

## 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

- Vertebrata
- Molluscs
  - -Malacothermometer
  - $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - $\delta^{18}O(^{16}O/^{18}O)$ 
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments

## **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology "Absolute" methods or numerical age estimation 10. Non-stable (radioactive) isotopes 10/a Potassium-argon series <sup>40</sup>K/<sup>40</sup>Ar 40Ar/39Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series <sup>238</sup>U/<sup>230</sup>Th <sup>234</sup>U/<sup>230</sup>Th <sup>230</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (238U) 10/d Radiocarbon dating (<sup>14</sup>C) 11. Luminescence methods -Thermoluminescence (TL) -Optically stimulated luminescence (OSL) InfraRed Stimulated Luminescence (IRSL) 12. Electron-spin resonance (ESR) 13. Cosmogenic isotopes

## **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

## 1. Lithostratigraphy

marker horizons:

tephrostratigraphy



# The importance of the marker horizons (in this case tephra layer) in Quaternary research in Hungary

• Projection of stratigraphical results from well investigated profiles to less known profiles

• Possibility of correlation with the stratigraphy of neighboring countries (eg. Bohemian, Slovakian, Croatian)

• Reconstruction of the paleosurface





#### Loess profiles with Bag Tephra layer in the Carpathian Basin



## **Bag Tephra in the Carpathian Basin**

Correlation based on:

- -mineralogical investigations,
- -grainsize distribution of clinopyroxenes,
- -geochemical investigations

## Most probable source area:

Appenninesacidic (Rome)<br/>semi acidic (Campania) lavasMto Valtano (Itala) hish K contact

Mte Vulture (Italy) high K-content

Active volcanoes in the time range 100-800ka: Massif Central, Aegean island arch, East-Carpathians, West-Carpathians, Volcanic Eifel



http://www.uni-miskolc.hu/~ecodobos/ktmcd1/huntalajok.htm









## **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

## 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

- Vertebrata fauna
- Molluscs
  - -Malacothermometer
  - $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - $\delta^{18}O(^{16}O/^{18}O)$ 
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments

## Vertebrata fauna in Hungary in Late-Pleistocene



9

## Gophers, "the foraminiferans of the Quaternary"...

## ... and other rodents

Hey lemming (Dicrostonyx torquatus)









Siberian gopher(Microtus gregalis)

Hamster (Cricetus cricetus)



ground-squirrel (Citellus citellus)

## Gophers in different environments



11

## Mammoth-stratigraphy



## **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

## 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

- Vertebrata fauna
- Molluscs
- -Malacothermometer
- $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - $\delta^{18}O(^{16}O/^{18}O)$ 
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments

Most common snails in the loess-paleosol sequences of the Carpathian Basin





15 Bösken et al. 2017.



## **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

## 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

- Vertebrata
- Molluscs
  - -Malacothermometer
  - $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - $\delta^{18}O(^{16}O/^{18}O)$ 
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments

Marine microfauna Foraminiferans



https://paleonerdish.wordpress.com/2013/06/17/an-introduction-to-foraminifera/

18

## **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

## 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

- Vertebrata
- Molluscs
  - -Malacothermometer
  - $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - $\delta^{18}O(^{16}O/^{18}O)$ 
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments

## Pollens



⊣ 50µm





Source: Sporomex

⊣ 10µm

## **Pollen diagrams**



## **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

## 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

- Vertebrata
- Molluscs
  - -Malacothermometer
  - $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - $\delta^{18}O(^{16}O/^{18}O)$ 
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments

## Biomarkers



## **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

## 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

- Vertebrata
- Molluscs
  - -Malacothermometer
  - $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - $\delta^{18}O(^{16}O/^{18}O)$ 
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments

# 3.1 Introduction **3. Magnetostratigraphy**

#### Origin of the Earth's geomagnetic field

The geomagnetic field is generated in some poorly understood way by the motion of highly conducting nickel-iron (Ni-Fe) fluids in the outer part of Earth' core; this motion is assumed to be controlled by thermal convection and by the Coriolis force generated by Earth's rotation, constituting what is called a **self-exciting dynamo**.



#### The Earth's magnetic field

When Earth's magnetic field has the present orientation, it is said to have **normal polarity**. When this orientation changes 180°, it has **reversed polarity**. Studies of the remanent magnetism in igneous and sedimentary rocks show that the dipole (main) component of Earth's magnetic field has reversed its polarity at irregular intervals from Precambrian time onward, apparently owing to instabilities in outer-core convection.

Reversals of Earth's magnetic field are recorded in sediments and igneous rocks by patterns of normal and reversed remanent magnetism. These geomagnetic reversals are contemporaneous worldwide phenomena. Thus, they provide unique stratigraphic markers in igneous and sedimentary rocks. The reversal process is thought to take place over a period of 1,000 – 10,000 years.

#### Docsity.com



## Magnetic susceptibility (MS)





## Magnetic susceptibility:

## $\mathbf{M} = \boldsymbol{\mu}_0 \boldsymbol{\kappa} \mathbf{H}$

**M**-magnetization, (magnetic dipole moment per unit volume); **H**-magnetic field strength;  $\mu_0$ -magnetic permeability;  $\kappa$ -Magnetic susceptibility, MS.

Magnetic susceptibility is a dimensionless proportionality constant that indicates the degree of <u>magnetization</u> of a material in response to an applied <u>magnetic field</u>.

If the magnetic susceptibility is greater than zero, the substance is said to be "paramagnetic"; the magnetization of the substance is higher than that of empty space. If the magnetic susceptibility is less than zero, the substance is "diamagnetic"; it tends to exclude a magnetic field from its interior. Mathematically it is the ratio of

magnetization **M** (magnetic moment per unit volume) to the applied magnetizing field intensity **H**.





## Forming and influencing factors of magnetization

•Ferromagnetic minerals: magnetite ( $Fe_3O_4$ ), maghemite ( $\gamma$ -Fe<sub>2</sub>O<sub>3</sub>), hematite (Fe<sub>2</sub>O<sub>3</sub>) and goethite ( $\alpha$ -FeO(OH));

•Paramagnetic minerals: clay minerals containing Fe<sup>2+</sup>, Fe<sup>3+</sup>, Mn<sup>2+</sup> ions, carbonates (siderite, rhodochrosite).

**paleosol** ↔ **loess: 2-3-fold difference!** 

## "Dilution hypothesis" (Kukla, G. et al. 1988):

Continuous sedimentation of the magnetic minerals (dust) on global scale (more or less independently from climate changes)

On local scale simultaneous sedimentation of coarser silt fraction from adjacent areas (the main volume of the loess), dilution of magnetic minerals.

"Soilformation hypothesis" (Heller, F. és Liu, T. S. 1986): formation of magnetic minerals due to soil forming processes.

## Magnetic susceptibility in loess-paleosol series



## **Discovery of tephra layers with MS investigations**



## Correlation of MS curves



## **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

## 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

- Vertebrata fauna
- Molluscs
  - -Malacothermometer
  - $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - $\delta^{18}O(^{16}O/^{18}O)$ 
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments

## Amino Acid geochronology



Leucozonella cf. rufispira (Martens 1874)

## Amino Acid geochronology

- relative age
- stratigraphical correlation

with calibration and/or the kinethic modell of racemization:

- numerical age estimation
- paleotemperature



# Influencing environmental factors of amino acid racemization


## **European loess-snails used in AAR studies**



#### DATING METHODS IN THE QUATERNARY STUDIES

#### **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

#### 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

#### 2. Biostratigraphy

- Vertebrata
- Molluscs
  - -Malacothermometer
  - $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - δ<sup>18</sup>O (<sup>16</sup>O/<sup>18</sup>O)
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments

# Isotopes

- Isotopes are distinct nuclear species (or <u>nuclides</u>, as technical term) of the same element. They have the same <u>atomic number</u> (number of <u>protons</u> in their <u>nuclei</u>) and position in the <u>periodic table</u> (and hence belong to the same <u>chemical element</u>), but differ in <u>nucleon</u> numbers (<u>mass numbers</u>) due to different numbers of <u>neutrons</u> in their nuclei.
  - For example, <u>carbon-12</u>, <u>carbon-13</u>, and <u>carbon-14</u> are three isotopes of the element <u>carbon</u> with mass numbers 12, 13, and 14, respectively. The atomic number of carbon is 6, which means that every carbon atom has 6 protons so that the neutron numbers of these isotopes are 6, 7, and 8 respectively.
  - A <u>nuclide</u> is a species of an atom with a specific number of protons and neutrons in the nucleus, for example carbon-13 with 6 protons and 7 neutrons. The *nuclide* concept (referring to individual nuclear species) emphasizes nuclear properties over chemical properties, whereas the *isotope* concept (grouping all atoms of each element) emphasizes <u>chemical</u> over nuclear.

#### Stable isotopes

- Their amount is constant in the earth's system
- Their ratio variable in the subsystems isotope fractionation (=the lighter isotope participate more likely in chemical, physical, biological processes than the heavier isotope of the same element)



## **Stable isotopes**



The **Oxygen-18 isotope** has an extra two neutrons, for a total of 10 neutrons and 8 protons, compared to the 8 neutrons and 8 protons in a normal oxygen atom. The *slighty greater mass of <sup>18</sup>O*—12.5 percent more than <sup>16</sup>O—results in differentiation of the isotopes in the Earth's atmosphere and hydrosphere. Scientists measure differences in oxygen isotope concentrations to **reveal past climates**. [Roll mouse over nuclei to animate.]

(Illustration by Robert Simmon, NASA GSFC)

$$\begin{split} \delta^{13}\mathrm{C} &= \left(\frac{\left(\frac{13C}{12C}\right)sample}{\left(\frac{13C}{12C}\right)standard} - 1\right) \times 1000\\ \delta^{18}\mathrm{O} &= \left(\frac{\left(\frac{13O}{16O}\right)sample}{\left(\frac{18O}{16O}\right)standard} - 1\right) \times 1000 \end{split}$$





https://earthobservatory.nasa.gov/Features/Paleoclimatology\_OxygenBalance/

Water vapor gradually loses <sup>18</sup>O as it travels from the equator to the poles. Because water molecules with heavy <sup>18</sup>O isotopes in them condense more easily than normal water molecules, air becomes progressively depleted in <sup>18</sup>O as it travels to high latitudes and becomes colder and drier. In turn, the snow that forms most glacial ice is also depleted in <sup>18</sup>O. As glacial ice melts, it returns <sup>16</sup>O-rich fresh water to the ocean. Therefore, oxygen isotopes preserved in ocean sediments provide evidence for past ice ages and records of salinity.

(Illustration by Robert Simmon, NASA GSFC)

#### **Importance of the unicellulars:**

## Oxygene isotope stratigraphy ( $\delta^{18}O$ )







Oxygen isotope curves of benthic foraminifera from the CSV-1 and KL-1 boreholes and a composite global deep-sea oxygen 45 isotope curve. Raw data for the global curve are based on benthic foraminifera from DSDP 77, DSDP 522, DSDP 529, DSDP 563, DSDP 574, ODP 689, ODP 744, ODP 748 and ODP 1218 sites (after Grossman, 2012).



https://timescavengers.blog/introductorymaterial/what-is-paleoclimatology/proxydata/carbon-oxygen-isotopes/







https://sciencestruck.com/isotopes-of-carbon



https://timescavengers.blog/introductory-material/what-ispaleoclimatology/proxy-data/carbon-oxygen-isotopes/

#### 2 different type of plants based on different photosynthetic pathway:

- C<sub>4</sub> plants: mild or cold growing season
  - Grasses and forbs
- C<sub>3</sub> plants: warmer and more humid environments
  - Distinction: closed canopy, top of the canopy, or open environment



		δ <sup>18</sup> Ο		δ <sup>13</sup> C	
		More Negative	More Positive	More Negative	More Positive
	Benthic oraminifera	-Warmer global climate -Decreased ice volume	-Colder global climate -Increased ice volume	-Older bottom waters -Increased continental	-Younger bottom waters -Decreased continental
	ш			weathering	weathering
	c era	-Warmer local temperatures	-Colder local temperatures	-Decreased local productivity	-Increased local productivity
	Plankti Foraminif	-Decreased ice volume	-Increased ice volume	-Increased upwelling (increased local	-Decreased-no upwelling (low local productivity)
		-Increased local precipitation	-Increased local evaporation	productivity)	·····



#### DATING METHODS IN THE QUATERNARY STUDIES

#### **RELATIVE DATING METHODS/RELATIVE STRATIGRAPHY**

#### 1. Lithostratigraphy

marker horizons:

tephrostratigraphy buried soils (paleosols, fossil soils)

#### 2. Biostratigraphy

- Vertebrata
- Molluscs
  - -Malacothermometer
  - $-\delta^{18}O$  of molluscs shells
- Marine microfauna
- Pollen
- Biomarkers
- 3. Magnetostratigraphy
- 4. Magnetic susceptibility (MS)
- 5. Amino Acid Racemisation (AAR)
- 6. Stable isotopes
  - $\delta^{18}O(^{16}O/^{18}O)$ 
    - (Milankovich theory)
  - $\delta^{13}C(^{12}C/^{13}C)$
- 7. Ice cores from the icesheets
- 8. Geomorphological horizons
  - terraces: fluvial marine
  - correlative sediments





Joseph Souney, Univ. New Hampshire (https://icecores.org/icecores/drilling.shtml)



A mechanical drill head showing the cutters used to shave an annulus around the ice to be cored. —Credit: Steven Profaizer, NSF



A thermal drill head showing the absence of cutters. Thermal drills use a heating element to melt an annulus around the ice to be cored. —Credit: Tony Wendricks, Univ. Wisconsin







## Investigation of ice cores

a. Melt Layers

cores taken in cold areas where melting infrequent frequency of melting = index of summer temperature

b. Gas Bubbles

gas composition set when ice forms from snow  $CO_2$  content in bubbles 100 ppm lower during Last Glacial Maximum

c. Annual Layer Thickness

Age determined by counting annual layers

Climate determined by annual layer thickness

d. Dissolved and Particulate Matter

Wind Blown Silt 3-20 times greater during isotope stage 2

Volcanic Ash also more abundant in stage 2

More dirt in Arctic cores = MELTED during Stage 5 (Koerner 1989)

Nitrate Ion: produced by UV radiation of  $N_2$  (solar)

e. Oxygen (and Hydrogen) Isotope Analysis Mirror image of ocean core isotopes: when ocean cores enriched in 18O, ice cores are depleted. Raleigh Fractionation Modern temperature-isotope relationships determined







300

https://timescavengers.blog/climate-change/co2past-present-future/



## Cooler and dustier climatic periods such as the Younger Dryas and Wisconsin glacial are characterized by high calcium concentrations in the GISP2 ice record.



61

Data in inset has not been averaged.

#### DATING METHODS IN THE QUATERNARY STUDIES

#### **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology "Absolute" methods or numerical age estimation 10. Non-stable (radioactive) isotopes 10/a Potassium-argon series <sup>40</sup>K/<sup>40</sup>Ar 40Ar/39Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series <sup>238</sup>U/<sup>230</sup>Th <sup>234</sup>U/<sup>230</sup>Th <sup>230</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (<sup>238</sup>U) 10/d Radiocarbon dating (<sup>14</sup>C) 11. Luminescence methods -Thermoluminescence (TL) -Optically stimulated luminescence (OSL) InfraRed Stimulated Luminescence (IRSL) 12. Electron-spin resonance (ESR) 13. Cosmogenic isotopes

## Varve chronology

**varved deposit** - any form of repetitive sedimentary rock stratification, either bed or lamination, that was deposited within a one-year time period.

This annual deposit may comprise paired contrasting laminations of alternately finer and coarser silt or clay, reflecting seasonal sedimentation (summer and winter) within the year.







http://eos.tufts.edu/varves/images/varve\_chron1.jpg



Abb. 10a und b: Ein Bohrkern mit klastischen Warven aus dem Holzmaar, einem Kratersee in der Vulkaneifel; links (a) ein Foto des erbohrten Kerns, rechts (b) ein Dünnschliff unter dem Mikroskop. Gröbere Mineralkörner zeigen die Phase des Schmelzwassereintrags an (Sommerlage). Davon deutlich abgetrennt bildet sich am Ende dieser Ablagerungsphase eine Lage aus feinem Ton, die sich erst unter den Stillwasserbedingungen des wieder zugefrorenen Sees absetzen kann. (Foto: Achim Brauer, GFZ)

#### Thinsection of a varve sediment



Coarser material transported by ice during spring



Abb. 12a bis c: Der typische Aufbau biogen-klastischer Warven am Beipiel von Jahreslagen im Holzmaar aus der kleinen Eiszeit; links (a) ein Foto des dem Seegrund entnommenen Sedimentbohrkerns, in der Mitte (b) die Dünnschliffaufnahme unter dem Mikroskop und rechts (c) der schematische Aufbau einer kompletten Jahreslage. (Fotos: Markus Schwab (a), Cathrin Brüchmann (b), Ulrike Krienel (c) GFZ)

## Stalagmite-chronology



## Dendrochronology



Scientists build tree-ring chronologies by starting with living trees and then finding progressively older specimens—including archaeological wood—whose outer rings overlap with the inner rings of more-recent specimens.





# A case study: joint application of tree-ring stable carbon and oxygen ( $\delta$ 13C and $\delta$ 18O) isotopes

## Recent European drought extremes beyond Common Era background variability

Ulf Büntgen, Otmar Urban, Paul J. Krusic, Michal Rybníček, Tomáš Kolář, Tomáš Kyncl, Alexander Ač, Eva Koňasová, Josef Čáslavský, Jan Esper, Sebastian Wagner, Matthias Saurer, Willy Tegel, Petr Dobrovolný, Paolo Cherubini, Frederick Reinig & Miroslav Trnka

Nature Geoscience (2021)

https://www.nature.com/articles/s41561-021-00698-0?fbclid=IwAR1HEPRr6TVpjb\_WIRJgCXc6U9TXpLikf7FxqTTsGzSonaOm8OKnHmuKP9s

# Growth characteristics and temporal coverage of the central European oak stable isotope dataset.

From: Recent European drought extremes beyond Common Era background variabilit



**a**, Temporal distribution of 147 living, historical, archaeological and subfossil oaks (green bars). The photographs at the bottom show examples of archaeological remains, subfossil materials, historical constructions and living oaks, and the grey shading on the right refers to the industrial period during which anthropogenic fossil fuel emissions affect the isotopic composition of CO<sub>2</sub>. **b**, Annually resolved TRW (left),  $\delta^{18}$ O (middle) and  $\delta^{13}$ C (right) measurement series aligned by their cambial ages (series length). The mean TRW,  $\delta^{18}$ O and  $\delta^{13}$ C values are 1.6 mm, 27.8‰ and -24.4‰, respectively (Extended Data Fig. 2). **c**, Microscopic amplification of an oak core sample that shows a sequence of well-defined tree**72** rings. Non-pooled TRSI measurements were extracted exclusively from the latewood alpha cellulose (*TRW* - tree-ring width; *TRSI* - tree-ring stable isotope)
# Reconstructed central European summer variability over the past 2,110 years.

From: Recent European drought extremes beyond Common Era background variability



Reconstructed JJA scPDSI from 75 BCE to 2018 CE (Supplementary Data 2). The thick green curve is a 50 yr cubic smoothing spline of the annual values, and the red and blue circles show the 20 lowest and highest reconstructed values, respectively (Extended Data Fig. 6). The grey shading refers to the confidence limits after smoothing, and the dashed line represents the highly significant long-term drying trend (y = -0.0012x + 2.4561,  $R^2 = 0.1281$ ). (PDSI - Palmer Drought Severity Index, based on a supply-and-demand model of soil moisture;

scPDSI - self-calibratedPDSI; JJA - June, July and August; LALIA - Late Antique Little Ice Age)

https://www.nature.com/articles/s41561-021-00698-0/figures/4

## Temporal and spatial agreement between the oak stable isotopes and European summer drought.

From: Recent European drought extremes beyond Common Era background variability



a, Actual (grey; averaged over 49–50° N and 15–18° E) and reconstructed (green; compound TRSI) JJA scPDSI. AC1 of the proxy and target data is provided at bottom left, whereas *r* values between the proxy and target data over the full and two early/late split periods are provided at the top right. b, High-resolution, 0.5° spatial correlation coefficients (colour scale) between the TRSI proxy data and the gridded European-wide scPDSI target data (left) and the regional average of the gridded scPDSI target data and European-wide gridded **74** scPDSI target data (right) over the common period 1901–2018 CE.

https://www.pature.com/articles/s41561-021-00698-0?fbclid=IwAR1HEPRr6TVpib\_WIR.JaCXc6U9TXpl.ikf7ExaTTsGzSonaOm8OKnHmuKP9s

#### **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology **"Absolute" methods or numerical age estimation** 10. Non-stable (radioactive) isotopes 10/a Potassium-argon series 40K/40Ar 40Ar/<sup>39</sup>Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series 2<sup>38</sup>U/<sup>230</sup>Th 2<sup>34</sup>U/<sup>230</sup>Th 2<sup>30</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (<sup>238</sup>U) 10/d Radiocarbon dating (<sup>14</sup>C)

#### Sir Edward Rutherford:

based on the decay time (half-lives) of radioactive elements, the age of the rocks can be determined

## Radioactive decay



#### **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology "Absolute" methods or numerical age estimation 10. Non-stable (radioactive) isotopes **10/a Potassium-argon series** 40 K/40 Ar40Ar/39Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series <sup>238</sup>U/<sup>230</sup>Th <sup>234</sup>U/<sup>230</sup>Th <sup>230</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (<sup>238</sup>U) 10/d Radiocarbon dating (<sup>14</sup>C) 11. Luminescence methods -Thermoluminescence (TL) -Optically stimulated luminescence (OSL) InfraRed Stimulated Luminescence (IRSL) 12. Electron-spin resonance (ESR) 13. Cosmogenic isotopes

## Potassium, Argon, and Calcium Rate Changes



ime

Argon 40 accumulates in the rock. Naturally occuring Ar 40 is assumed to have escaped when the rock was hot. Calcium 40 is produced in the rock but there is no way to make any measurements of the accumulating Ca 40. Naturally occuring Ca 40 would contaminate the measurements. Potassium 40 is used up in the rock. It breaks down into both Ar 40 & Ca 40.



• 10/a Potassium argon series

<sup>40</sup>K/<sup>40</sup>Ar <sup>40</sup>Ar/<sup>39</sup>Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar

\_\* <sup>40</sup>K → <sup>40</sup>Ar (T<sub>1/2</sub>=1,25 md year, 11,71% <sup>40</sup>Ar\*) cooling of rock → incorporation of atmospheric<sup>40</sup>Ar/<sup>36</sup>Ar → production of <sup>40</sup>Ar\* radioactive decay (time, K-content)
\* <sup>40</sup>Ar/<sup>39</sup>Ar (fast neutron irradiation <sup>39</sup>Ar → <sup>40</sup>Ar → advantage: only Ar measurment
\* laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar

> Gradual heating analytical age geological age

#### **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology "Absolute" methods or numerical age estimation 10. Non-stable (radioactive) isotopes 10/a Potassium-argon series  $^{40}K/^{40}Ar$ 40Ar/39Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series <sup>238</sup>U/<sup>230</sup>Th <sup>234</sup>U/<sup>230</sup>Th <sup>230</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (238U) 10/d Radiocarbon dating (<sup>14</sup>C) 11. Luminescence methods -Thermoluminescence (TL) -Optically stimulated luminescence (OSL) InfraRed Stimulated Luminescence (IRSL) 12. Electron-spin resonance (ESR) 13. Cosmogenic isotopes

## **Uranium-238 decay series**



## Uranium – thorium <sup>234</sup>U/<sup>230</sup>Th

- Age range:
  - -0-0.5 Ma
  - Material:

•

- Corals, cave deposits, pit, bone, mollusc's shell, tooth, travertines
- U soluble in water build into • carbonate
- Th insoluble--product of • decay of U = time
- Measurements:  $\alpha$  counting, • mass spectrometry



#### **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology "Absolute" methods or numerical age estimation 10. Non-stable (radioactive) isotopes 10/a Potassium-argon series  $^{40}K/^{40}Ar$ 40Ar/39Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series <sup>238</sup>U/<sup>230</sup>Th <sup>234</sup>U/<sup>230</sup>Th <sup>230</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (<sup>238</sup>U) 10/d Radiocarbon dating (<sup>14</sup>C) 11. Luminescence methods -Thermoluminescence (TL) -Optically stimulated luminescence (OSL) InfraRed Stimulated Luminescence (IRSL) 12. Electron-spin resonance (ESR)

13. Cosmogenic isotopes

## Fission Tracks:





## **Fission Tracks in crystal**





#### **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology "Absolute" methods or numerical age estimation 10. Non-stable (radioactive) isotopes 10/a Potassium-argon series  $^{40}K/^{40}Ar$ 40Ar/39Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series <sup>238</sup>U/<sup>230</sup>Th <sup>234</sup>U/<sup>230</sup>Th <sup>230</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (<sup>238</sup>U) **10/d Radiocarbon dating** (<sup>14</sup>C) 11. Luminescence methods -Thermoluminescence (TL)

-Optically stimulated luminescence (OSL)

InfraRed Stimulated Luminescence (IRSL)

12. Electron-spin resonance (ESR)

13. Cosmogenic isotopes



© Pass My Exan



The <sup>14</sup>C-content of the atmospheric carbon reservoir is constant! 87 (13,6 dpm / g carbon)



## Dating based on the measurment of <sup>14</sup>C amount (activity)



**Conventional radiocarbon ages:** 

$$t(y_{ear}) = \frac{5568}{\ln 2} * \ln\left(\frac{A_{begin}}{A}\right)$$

Most accurate half-lives of <sup>14</sup>C (when the theory is created – in 1950): 5568 years

Assumption: A  $_{\text{begin}}$  constant, atmospheric concentration of  $^{14}\text{C}$  did not change during time, the same as it was in 1950.

<u>Agreement:</u> all age is given relative to A.D.1950. Example: if the **conventional radiocarbon age** is **2500 years BP** (before present), the calendar time is A.D. 550 years:

1950-2500 = -550, **i.e. A.D. 550** ....

## **Materials to date:**

## **Everything containing atmospheric carbon.**

Wood, charcoal, seed, leaf, canvas, peat, humus, bone, ivory, cloth, horn, hair, shell, snail, carbonate, stalagtit, soil, organic and inorganic carbon dissolved in water, ice...

Usually not suitable for dating: plaster, mortar, organic material in ceramics **Suitable with special technique:** *paintings, cave paintings, carbon in iron tools*.

The amount of sample depends on:

Carbon content, the condition of the organic material, amount of contamination, and the method

Deviation from the BP age because of the different atmospheric radiocarbon content in the past





#### Changes in the <sup>14</sup>C concentration due to human impact

Suess-effect: use of the fossil fuel caused the increase of the inactive CO<sub>2</sub> concentration in the atmosphere until 1952: 3% decrease of the <sup>14</sup>C concentration









 $^{3}\mathbf{H}$ 







129<mark>1</mark>

# Amount of <sup>14</sup>C in the atmosphere due to the nuclear tests:



# The carbon content of tree ring represents the actual 14C concentration of the atmosphere



#### How radiocarbon calibration works

Calibration of radiocarbon determinations is in principle very simple. If you have a radiocarbon measurement on a sample, you can try to find a **tree ring with the same proportion of radiocarbon**. Since the calendar age of the tree rings is known, this then tells you the age of your sample. In practice this is complicated by two factors:

•the measurements on both the tree rings and the samples have a **limited precision** and so there will be a range of possible calendar years •given the way the atmospheric radiocarbon concentration has varied, there might be several possible ranges



The radiocarbon measurement **3000+-30BP would be calibrated**. The left-hand axis shows *radiocarbon concentration* expressed in years `before present' and the bottom axis shows *calendar years* (derived from the tree ring data). The pair of blue curves show the *radiocarbon measurements on the tree rings* (plus and minus one standard deviation) and the red curve on the left indicates the radiocarbon concentration in the sample. The grey histogram shows possible ages for the sample (the higher the histogram the more likely that age is).

The results of calibration are often given as an age range. In this case, we might say that we could be 95% sure that the sample comes from between 1375 cal BC and 1129 cal BC.

98



#### **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology "Absolute" methods or numerical age estimation 10. Non-stable (radioactive) isotopes 10/a Potassium-argon series  $^{40}K/^{40}Ar$ 40Ar/39Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series <sup>238</sup>U/<sup>230</sup>Th <sup>234</sup>U/<sup>230</sup>Th <sup>230</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (238U) 10/d Radiocarbon dating (<sup>14</sup>C) 11. Luminescence methods -Thermoluminescence (TL) -Optically stimulated luminescence (OSL) InfraRed Stimulated Luminescence (IRSL) 12. Electron-spin resonance (ESR)

13. Cosmogenic isotopes











## Luminescence dating

**Optical luminescence:** 

- IRSL: Infrared Stimulated Luminescence
   feldspar
- •BLSL: Blue Light Stimulated Luminescence quartz





Grain size:

Coarse grain (quartz or K-feldspar): 100-200 µm (sand)

Polymineral (fine grain): 4-11 µm (loess)

## Luminescence dating methods

- Thermoluminescence (TL)
- Optically Stimulated Luminescence(OSL)
  - Infrared (IRSL)
  - Red OSL
  - Green OSL
  - Blue OSL
  - Pulsed OSL
  - TT OSL (up to 1 My!)
- Infrared radiofluorescence (IR-RF)






## DATING METHODS IN THE QUATERNARY STUDIES

## **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology "Absolute" methods or numerical age estimation 10. Non-stable (radioactive) isotopes 10/a Potassium-argon series  $^{40}K/^{40}Ar$ 40Ar/39Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series <sup>238</sup>U/<sup>230</sup>Th <sup>234</sup>U/<sup>230</sup>Th <sup>230</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (238U) 10/d Radiocarbon dating (<sup>14</sup>C) 11. Luminescence methods -Thermoluminescence (TL) -Optically stimulated luminescence (OSL) InfraRed Stimulated Luminescence (IRSL) 12. Electron-spin resonance (ESR) 13. Cosmogenic isotopes

## DATING METHODS IN THE QUATERNARY STUDIES

## **ABSOLUTE DATING METHODS**

9. Varve-, and dendrochronology "Absolute" methods or numerical age estimation 10. Non-stable (radioactive) isotopes 10/a Potassium-argon series  $^{40}K/^{40}Ar$ 40Ar/39Ar laser fusion <sup>40</sup>Ar/<sup>39</sup>Ar 10/b Uranium decay series <sup>238</sup>U/<sup>230</sup>Th <sup>234</sup>U/<sup>230</sup>Th <sup>230</sup>Th/<sup>231</sup>Pa (<sup>235</sup>U) 10/c Fission-track (238U) 10/d Radiocarbon dating (<sup>14</sup>C) 11. Luminescence methods -Thermoluminescence (TL) -Optically stimulated luminescence (OSL) InfraRed Stimulated Luminescence (IRSL) 12. Electron-spin resonance (ESR)

13. Cosmogenic isotopes

# Age determination using terrestrial in-situ produced cosmogenic nuclides (TCN)

**Exposure age** of a rock is the time elapsed since it has been exposed to cosmic irradiation = since it is on the sur  $\int |N| = P_{\times t}$ 



## : TCN build-up

- <sup>3</sup>He; <sup>14</sup>C; <sup>10</sup>Be; <sup>21</sup>Ne; <sup>26</sup>Al; <sup>36</sup>Cl

Burial age is the time elapsed since a rock surface or sediment is shielded from cosmic irradiation.



: decay of radioactive TCN

2 nuclides of different half-lives are neccesary

- <sup>26</sup>Al/<sup>10</sup>Be; <sup>21</sup>Ne/<sup>10</sup>Be; <sup>14</sup>C/<sup>10</sup>Be





### Accumulation of cosmogenic nuclides

#### Prerequisites of reliable age information:

- Formation of the dated surface is instantaneous
- No erosion and no aggradation since formation (!?)

#### The effect of erosion or aggradation on the exposure age

Beneath the surface, production of cosmogenic nuclides decreases exponentially.



**Temporal cover** (loess, soil, snow) is shielding the surface from cosmic irradiation.

#### The effect of erosion on TCN concentration in time



Using surface samples: minimum exposure age or maximum erosion rate !

# Age range of the different dating methods in the Quaternary studies



# Loess – windows to the Past

A terrestrial climate archive from the Pleistocene



## LOESS RESEARCH

• Chinese Loess Plateau by J. Cholnoky

- First description of loess:
- Von Leonhard (Haarlass, near Heilderberg, Germany) 1824



#### 89: Loefs.

Syn. Loesch; Schneckenhäusel. Boden; Mergel (im Badist Oberlande); Briz.

STRININGER. Neue Beiträge sur Geschichte der Rheinischen Vulkane; 5: ff.

Mehrere Nouzen, den Löfs betreffend, sind entlehnt aus des II D' BRONN und des Verfassers Manuscript über das Groblahk-Gebilde Rheinthale.

Ein lehmiges, unrein gelblichgraues, im Bruc erdiges Gemenge aus Thon-, Kalk- und Kieselth len und aus sehr kleinen Glimmer-Blättchen. I feinstaubigen Theile sind zu einer losen zerreib chen Masse verbunden.

Im Löß der Heidelberger Gegend finden sich, auf 2 The Thon; 1/2 Theil kohlensaurer Kalk und 1/2 Theil quarziger und gli meriger Sand.

In der Gegend von Andernach trifft man, jedoch nur sp sam, auch Bimsstein - und Schlacken-Stückchen im Löfs.

Eine, über dem Löß liegende, Schicht von neuerer Ent hung, untermengt mit Kalk-Geröllen, umschließt einige Ar von Bulimus und von Helix, welche in der nachbarlichen ( gend noch heutiges Tages in größter Menge vorkommen.

Schlofsberg bei Oppenheim.

Der Löfs enthält versteinte und kalzinirte Ko chylien.

Viele sind, vom Tage her, in seine Massen gefl worden; als ihm mehr eigenthümlich zugehörig, wurden jezt nur wenige beobachtet, so namentlich einige Hel. und Lymnaeus-Arten.

Die größern u. a. bei Oppenheim am Galgenberge, die kleinern Weinheim unfern Alsey, am Haarlafs und bei Neckargemund.

Aus dem Löfs der Gegend von Andernach führt Strimingen an: H. pomatia, nemoralis, hortensis, striata, pulchella und cristallina; L naeus pereger und ovatus; Pupa edentula.

Die Schaalen der Muscheln tragen zum Theil noch i natürliche Farbe; die meisten sind jedoch weiß und e zerbrechlich.

Auch Stofs- und Mahlzähne vom Mammuth wurd nebst andern, nicht näher bestimmten, meist sehr au: lösten Knochen im Löß gefunden.

So u. a. unfern Weinheim in der Bergstrafse.

Auf Veranlausang und durch Vorsorge der Herren v. BABO and Dr. BATT m & Auf Verstalsung und durch vonorfe der fieren v. SABO und U. Sall im / Ann. weiere ben warmen als einschlutellen. Matterfrande, write dies interträtt im frählinge idad durch Wauereisse in einem Hohlwege enthlöhten, kolosalen Reise writtere Zenstörang geschlutt. Sie befinden sich jest in den Sommlangen der Universi Heidelberg.

Der Löß gibt, bei gehöriger Behandlung mit Dünger, ei: rumal für Weinbau, sehr diensamen Boden.



# Why loess research is important in Hungary?

- Archive of the Quaternary climate and environmental changes
- Importance of the Carpathian Basin in the West-East loess-transect
- Tradition (thick loess, outcrops, brickyards)
- Big "chaos" in:
  - Number of buried soils
  - Numbering or labeling of paleosoils
  - Position of the last interglacial

# History of Loess research in Hungary

- Szabó, J. (1877): sea or lake sediment
- Inkey, B. (1878), Halaváts, Gy. (1895), Treitz, P. (1901), Horusitzky, H. (1898, 1903): subaerial sediment
- > Cholnoky J. (1902): dust from Inner-Asia
- > Prinz Gy. (1926): dust from the margin of the Northern-European icesheet
- Bulla, B. (1934, 1938):
  - > Loess: glacials, loamy horizons (=paleosoils): interglacials (according to Soergel's theory)
  - > Dust principally from "floodplains of the Hungarian-Basin"
- Pécsi, M. (1969, 1975):
  - > Poligenetic origin

(frost weathering in mountains  $\rightarrow$  mass movement  $\rightarrow$  fluvial transport  $\rightarrow$  accumulation on alluvial fans and floodplains  $\rightarrow$  eolian transport  $\rightarrow$  dust accumulation on river terraces, hilly regions, flat surfaces  $\rightarrow$  diagenesis)



# paleosol

Carbonate accumulation

## paleosol



Ice age=glacial, stadial



Warmer period=interglacial, interstadial

#### Polygenetic theory of loess formation



#### CaCO<sub>3</sub> in the matrix - the **diagenesis of the loess**



Diagenesis after dust accumulation (from sediment to loose clastic sedimentary rock)

## Loess on terraces in Hungary



# Correlation of the theoretical loess-paleosol sequence (based on the results of different Hungarian loess outcrops) with the marine oxygen isotope stages













## Multiproxy studies



Luminescence dating <sup>14</sup> C • • AAR Malakology Magnetic susceptibility (5cm) Paleomagnetic • investigations (10cm) Granulometry (2cm) 00 Stabile isotopes (2cm) • 0 Geochemical investigations • • (2cm) Biomarkers (25cm) Secondary carbonates Thinsections (selected)

O.



# Multiproxy studies 2. (grainsize, magnetic susceptibility

Multiproxy studies 3.  $(CaCO_3)$ 



## Tephrostratigraphy





(Horváth 2001, Horváth et al. 2006)

## The importance of the marker horizons (here: tephra layer) in Quaternary research in Hungary

Projection of stratigraphical results from well investigated profiles to less known profiles
Possibility of correlation with the stratigraphy of neighbouring countries (eg. Bohemian, Slovakian, Croatian)
Reconstruction of the paleosurface



The Bag Tephra layer in the outcrops and in thinsection



D 80 20 Ö



Reconstruction of a paleosurface based on the Bag tephra layer in the outcrop of Isaszeg – an example of the geomorphological inversion



The volcanic ash deposited in a valley

today it is a loessridge between two valleys





In: Horváth E, Bradák B. Sárga föld, lősz, lösz: short historical overview of loess research and lithostratigraphy in Hungary. QI 319:(1) pp. 1-10. (2014)

	<b>Bulla, 1938</b>	Kriván,	Pécsi, 1975	Wintle-	Oches-McCoy,	Pécsi,	Frechen et al., 1997,
		1955		Packman, 1988	1995	1995	Novothny et al. 2002
			<i>C</i> <sup>14</sup>	TL	AAR	$C^{14}, TL,$	TL+IRSL
						(estimation)	
rec. talaj						0-11,3	
l1				15,8±1,3			
l1				17,6±1,4			(13-20)
h1			(16,75±0,4)			<b>W3</b> (16-17)	
l1				23,2±1,9			
h2			(20-22)			W2/W3 (27-32)	
12				24,0±2,0			(25-35)
MF1			(28-29)	20,9±1,7		<b>W2/W3</b> (45-60)	
13				43,4±3,8			(50-60)
MF2			(32)	<b>R/W</b> (74,7±6,5)	<b>MIS 5</b> (<128 BP)	<b>W1/W2</b> (85-105)	MIS 5 (65±10)
14		W2		<b>R</b> (79,2±6,1)			<b>MIS 6</b> (>130)
BD1	W2/W3	W2				<b>R2/W1</b> (120-140)	
		W1					
BD2	W1/W2				<245 BP	<b>R2/W1</b> (150-170)	
15		W1					
BA	R/W				<339 BP	<b>R1/R2</b> (195-230)	
<b>l6</b>		R2					
MB1			R/W			<b>MR3/MR4</b> (280-310)	
MB2			R/W		<423 BP	<b>MR2/MR3</b> (320-360)	
L1		R1					

## Chistory of the chronostratigraphy of the Hungarian loess (ka)

(After Horváth 2001)







Powdery accumulation of CaCO<sub>3</sub>



Biospheroid (EBS)

## Accumulation of CaCO<sub>3</sub> in different forms: Secondary carbonates



CaCO<sub>3</sub> Infilling of cracks



CaCO<sub>3</sub> concretion (loessdoll)





#### **Calcified rootcells**











Secondary carbonates:

Calcified rootcells - reworked



## Secondary carbonates



## Needle fiber calcite





 HV
 Spot
 Det
 Mag
 WD
 HFW
 8/9/2010

 20.0 kV
 5.0
 SSD
 1200x
 10.3 mm
 0.23 mm
 3:31:18 PM



Secondary carbonates: Earhtworm "calculus": Biospheroid

• See more about secondary carbonates: **Barta, G.** 2014. Paleoenvironmental reconstruction based on the morphology and distribution of secondary carbonates of the loesspaleosol sequence at Süttő, Hungary. *Quaternary International 319*, pp. 65-75.
# Secondary carbonates

## HC - Hypocoating



CaCO<sub>3</sub> coating (CCo - coating)





Similar features in different climate: semiarid climate (Iran)

# Gypsum instead of CaCO<sub>3</sub>



# Clay coating



Smooth and shiny surfaces, usually on ped surfaces, but also along biogalleries



Clay and Mn coating

Mixed features

Clay coating on CaCO<sub>3</sub> coating



## Different forms of clay in the loess-paleosol series





#### Signs of reworking in loess (in thin section)





• Reworking by geomorphological processes

## Landslide on loess – Dunaszekcső (Hungary)



# Signs of freezingthawing





# Bioturbation







vertical

horizontal

## Recent bioturbation

On the surface







## **Recent bioturbation**



Insects holes in loess



## Charcoal in loess















# Wetting-drying





# Wetting-drying







# Hydromorphic features



Mn on clay coating





Fe along roots



Mn concretions

901

188

186

030181



# Hydromorphic features

Recent environment (small intermittent lake in sandy area)









# • Quaternary in the Carpathian Basin

# Different environments during the glacials and intergalcials







# **Quaternary in Hungary**



## **Quaternary surface development in the Carpathian Basin**



Environments during the glacials\*

- High mountains (Alps, Carpathians)
- Lower mountains (in the Carpathian Basin)
- Hilly regions
- Lower part of the basin (floodplains, alluvial fans, subsiding basins)

\*The environments during interglacials are similar to the recent environment without anthropogenic impact

#### High mountains (Alps, Carpathians):

- Glaciers
- Nivation
- frost weathering





#### The landscape after the deglaciation





- Lower mountains (in the Carpathian Basin)
- frost and insolation weathering
- Rockfalls
- Talus development



#### Loess on travertine (calc sinter)



- Hilly regions and river terraces (hills and piedmont surfaces):
- Mass movement
- Pediment formation
- Dust accumulation and loess formation

## Contact point of the mountains and the basin



shutterstock.com • 271732646



Dominant Geomorphic Processes:

Landform Scale (Order 2):







Cryoturbation in terrace sediments





#### Cryoturbation in loose sediments



#### A sand dune and its cross section





- Lower part of the basin (floodplains, alluvial fans, subsiding basins):
- Eolian sand accumulation
- Accumulation in the riverbed
- Sediment accumulation in subsiding basins
- Cryoturbation in sediments
- Permafrost?



Quaternary alluvial fans of the Great Plain of Hungar Inferred sediment transport direction and grain size (after Borsy 1992)





High-resolution stratigraphy of a Quaternary fluvial deposit based on magnetic susceptibility variations (Jászság Basin, Hungary)

Boreas, Volume: 49, Issue: 1, Pages: 181-199, First published: 27 September 2019, DOI: (10.1111/bor.12412)

High-resolution stratigraphy of a Quaternary fluvial deposit based on magnetic susceptibility variations (Jászság Basin, Hungary)



Boreas, Volume: 49, Issue: 1, Pages: 181-199, First published: 27 September 2019, DOI: (10.1111/bor.12412)




## River terraces – the results of the consecutive climate changes





Castle Hill, Buda - IV. Terrace of the river Danube



## The Hungarian terrace system

- T = termination
- IS = interstadial
- MIS = marine oxygenisotope stadium
- ka = 1000 years



